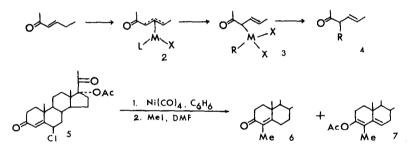
ALKYLATION OF ENONES VIA T-ALLYL COMPLEXES

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Syntex Research, 3401 Hillview Ave., Palo Alto, California 94304 (Received in USA 17 February 1969; received in UK for publication 24 March 1969) A recent publication¹ describing the preparation and reactions of PdCl₂ π -allyl complexes of steroids prompts us to present our work in this field. The potential utility of π -allyl complexes as reactive intermediates for the alkylations of enones, under mild non-basic conditions, is suggested by the known coupling of π -allyl NiBr₂ with alkyl halides.² Thus reaction of the complex 2 with alkyl halides would be expected to furnish the alkylated enone 4, conceivably by way of an intermediate σ -complex 3.²

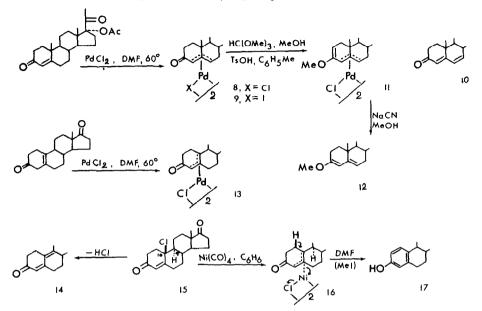


Treatment of the enone 5 with nickel carbonyl gave a red complex which reacted with methyl iodide forming the 4-alkylated derivatives 6 and 7 in about 24% combined yield. The enol acetate 7 presumably arises from a reaction of the steroid with acetyl iodide, a carbonylation product of methyl iodide.

Palladium chloride n-allyl complexes on the other hand are less reactive than their nickel analogs. Predictably, the complex 8 was not alkylated by methyl iodide although the previously noted halogen exchange,¹ forming the palladium iodide complex 9, was observed. Reaction of the complex 8 with the strong nucleophiles cyanide and malonate ion led to the elimination product 10. On treatment with methyl orthoformate 8 gave the enol ether 11 which was reduced to 12 by sodium cyanide in methanol. The novel PdCl₂ complexes 8. 11 and 13 were prepared directly from the steroid by heating with PdCl₂ in DMF.³

Attempted alkylation of the NiCl $_2\pi$ -complex 16 gave a good yield of estrone

17. It has been shown⁴ that base treatment of 10-chloro-enones similar to 15 normally leads to the dienone 14 by a trans-elimination of HCl. The formation of estrone in our case suggests the α -stereochemistry of the NiCl₂ moiety in 16 which would be trans with respect to the β -hydrogen on C-1.



Complex	{α} _D	λ max (e) (mμ)	δ vinyl <u>2-H</u>	protons (ppm) <u>4-H</u>	<u>6-H</u>
8	+260 $^\circ$ (diox.)			3.43(s)	4.38(m)
9	+183° (chl.)	278 (16,100)(meth.)		3.52(s)	4.60(m)
11	-632 $^{\circ}$ (diox.)		5.07(m)	3.65(s)	4.09(d)
13	+226 $^{\circ}$ (chl.)	280 (11,000) (chl.)		4.76(s)	

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